UNUSUAL DESHIELDING OF <sup>15</sup>N IN BICYCLIC AMINES AND HYDRAZINES

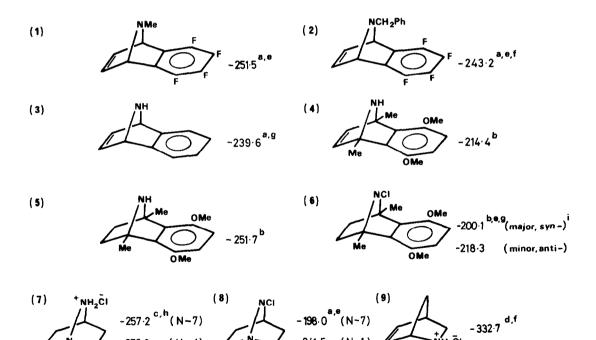
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Summary: The bridging nitrogen in 7-aza- and 1,7-diazabicyclo[2,2.1]heptanes and unsaturated derivatives shows unusually large deshielding in the <sup>15</sup>N NMR.

The bridging nitrogen in 7-azabicyclo[2.2.1]heptanes and derivatives is unusual, showing remarkably high barriers to inversion at nitrogen  $^1$  and some intriguing chemistry. $^2$  We have measured <sup>15</sup> N NMR spectra for a variety of N-bridged systems and find that these amines display the lowestfield  ${}^{15}_{N}$  signals yet recorded for secondary and tertiary amines (Table).

<sup>15</sup>N Chemical Shifts (relative to CH<sub>3</sub>NO<sub>2</sub>) TABLE



(a)  $CD_2Cl_2$  (b)  $CD_2Cl_2/CH_2Cl_2$  (c)  $D_2O$  (d)  $H_2O/D_2O$  (e) with added  $Cr(acac)_3$  (f) with gated decoupling (g) with no decoupling (h) with inverse gating, no NOE (i)  $Cl \underline{syn}$ - to dimethoxybenzo-ring. [Natural abundance  ${}^{15}N$  spectra measured at 40.55 MHz.]

Thus nitrogen falls into line with oxygen,<sup>3</sup> phosphorus,<sup>4</sup> and silicon<sup>5</sup> which all show unusual deshielding in the 7-position of bicyclo[2.2.1]heptane, heptene or heptadiene skeleta.

The <sup>15</sup>N shifts for the tertiary amines (1) and (2) are over 70 ppm downfield of the most deshielded amine nitrogen previously recorded.<sup>6</sup> The secondary amines (3), (4), and (5) also show large downfield shifts and the downfield increment of <u>ca</u>. 37 ppm on replacing an ethanobridge with an etheno- bridge [(5) + (4)] parallels a similar trend in C,<sup>3</sup>P,<sup>4b</sup> and o<sup>3</sup> systems. The N-chloroamine (6) is interesting for two reasons: firstly, it provides the lowest-field resonance for an amine nitrogen in this study<sup>7</sup> and secondly, it constitutes the first example of a system showing separate signals for the two invertomers at a single nitrogen.<sup>8</sup> Integration of the <sup>15</sup>N spectrum gave a ratio of 65:35 which corresponds closely with the (more reliable) ratio of 71:29 which was measured by <sup>1</sup>H nmr and assigned to the invertomers having the chlorine <u>syn</u>-and <u>anti</u>- to the aryl ring, respectively.

In the bicyclic hydrazine (7), N-1 is just within the 'normal' range for hydrazines (-270 to  $-330 \text{ ppm}^6$ ); N-7 is deshielded. The N-chloro- analogue (8) shows significant deshielding of both nitrogens which suggests that the inductive effect of Cl is a factor. The bicyclic amine (9) is included as a point of reference and appears at an unexceptional chemical shift, emphasising the special nature of N-7 in this ring system.

It is unlikely that angle strain is the major factor since the CNC bond angle in these systems is less than that in azetidines<sup>1</sup> and, in any event, nitrogen in strained rings is normally highly shielded (e.g. aziridine,  $\delta$  -393.3<sup>6</sup>). The deshielding of nitrogen in these bridged amines is consistent with increased delocalisation of the nitrogen lone pair when compared with normal amines.<sup>9</sup> We intend to extend the range of bridged amines and investigate possible empirical correlations of <sup>15</sup>N chemical shifts with (i) specific structural features, (ii) chemical shifts of other bridging atoms in similar systems and (iii) inversion barriers at nitrogen.

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   We have, as yet, no data on the N-chloro-derivative of (4) which might have been expected to show even more deshielding of the nitrogen due to the presence of chlorine and the etheno-
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