

UNUSUAL DESHIELDING OF ^{15}N IN BICYCLIC AMINES AND HYDRAZINES

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Summary: The bridging nitrogen in 7-aza- and 1,7-diazabicyclo[2.2.1]heptanes and unsaturated derivatives shows unusually large deshielding in the ^{15}N NMR.

The bridging nitrogen in 7-azabicyclo[2.2.1]heptanes and derivatives is unusual, showing remarkably high barriers to inversion at nitrogen¹ and some intriguing chemistry.² We have measured ^{15}N NMR spectra for a variety of N-bridged systems and find that these amines display the lowest-field ^{15}N signals yet recorded for secondary and tertiary amines (Table).

TABLE ^{15}N Chemical Shifts (relative to CH_3NO_2)

(1)		-251.5 ^{a,e}
(2)		-243.2 ^{a,e,f}
(3)		-239.6 ^{a,g}
(4)		-214.4 ^b
(5)		-251.7 ^b
(6)		-200.1 ^{b,e,g} (major, syn-) ⁱ -218.3 (minor, anti-)
(7)		-257.2 ^{c,h} (N-7) -276.0 (N-1)
(8)		-198.0 ^{a,e} (N-7) -241.5 (N-1)
(9)		-332.7 ^{d,f}

(a) CD_2Cl_2 (b) $\text{CD}_2\text{Cl}_2/\text{CH}_2\text{Cl}_2$ (c) D_2O (d) $\text{H}_2\text{O}/\text{D}_2\text{O}$ (e) with added $\text{Cr}(\text{acac})_3$ (f) with gated decoupling (g) with no decoupling (h) with inverse gating, no NOE (i) Cl *syn*- to dimethoxybenzo- ring. [Natural abundance ^{15}N spectra measured at 40.55 MHz.]

Thus nitrogen falls into line with oxygen,³ phosphorus,⁴ and silicon⁵ which all show unusual deshielding in the 7-position of bicyclo[2.2.1]heptane, heptene or heptadiene skeleta.

The ¹⁵N shifts for the tertiary amines (1) and (2) are over 70 ppm downfield of the most deshielded amine nitrogen previously recorded.⁶ The secondary amines (3), (4), and (5) also show large downfield shifts and the downfield increment of ca. 37 ppm on replacing an ethano-bridge with an etheno- bridge [(5)→(4)] parallels a similar trend in C,³ P,^{4b} and O³ systems. The N-chloroamine (6) is interesting for two reasons: firstly, it provides the lowest-field resonance for an amine nitrogen in this study⁷ and secondly, it constitutes the first example of a system showing separate signals for the two invertomers at a single nitrogen.⁸ Integration of the ¹⁵N spectrum gave a ratio of 65:35 which corresponds closely with the (more reliable) ratio of 71:29 which was measured by ¹H nmr and assigned to the invertomers having the chlorine syn- and anti- to the aryl ring, respectively.

In the bicyclic hydrazine (7), N-1 is just within the 'normal' range for hydrazines (-270 to -330 ppm⁶); N-7 is deshielded. The N-chloro- analogue (8) shows significant deshielding of both nitrogens which suggests that the inductive effect of Cl is a factor. The bicyclic amine (9) is included as a point of reference and appears at an unexceptional chemical shift, emphasising the special nature of N-7 in this ring system.

It is unlikely that angle strain is the major factor since the CNC bond angle in these systems is less than that in azetidines¹ and, in any event, nitrogen in strained rings is normally highly shielded (e.g. aziridine, δ -393.3⁶). The deshielding of nitrogen in these bridged amines is consistent with increased delocalisation of the nitrogen lone pair when compared with normal amines.⁹ We intend to extend the range of bridged amines and investigate possible empirical correlations of ¹⁵N chemical shifts with (i) specific structural features, (ii) chemical shifts of other bridging atoms in similar systems and (iii) inversion barriers at nitrogen.

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7. We have, as yet, no data on the N-chloro-derivative of (4) which might have been expected to show even more deshielding of the nitrogen due to the presence of chlorine and the etheno-bridge; such N-chloroamines are highly reactive.
8. Previously reported examples have involved double inversion in bicyclic hydrazines; Y. Nomura and Y. Takeuchi, *J. Chem. Soc. Chem. Comm.*, 1979, 295.
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